# FLAVONOL DERIVATIVES OF DESMANTHODIUM (COMPOSITAE)

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(Received 11 November 1980)

**Key Word Index**—Desmanthodium; Compositae; Heliantheae; Milleriinae; flavonols; O-methylation; chemosystematics.

Abstract—The flavonoids of three species of *Desmanthodium* are based upon kaempferol, quercetin and quercetagetin. Sugar substitutions comprise glucosides, galactosides, rhamnosides, rutinosides and diglucosides. Four different Omethylated compounds occur in field populations of the genus, but they are found in all species and are therefore not useful for sectional or subgeneric delimitations. The flavonoid profile of *Desmanthodium* is very similar to that of *Clibadium*, which parallels their close morphological affinity.

#### INTRODUCTION

As a continuation of our biosystematic study of the subtribe Milleriinae of the Heliantheae, we undertook an examination of the flavonoid chemistry of three species of Desmanthodium Benth: D. fruticosum Greenm., D. hondurense Molina and D. perfoliatum Benth. The genus consists of ten species and is known from Mexico and Central America [1]. A recent communication [2] described our study of the flavonoids of the closely related genus Clibadium. To the best of our knowledge, no other work on the flavonoids of Desmanthodium has been described.

### RESULTS

The flavonoids of the *Desmanthodium* species examined in this study were all based upon kaempferol, quercetin and quercetagetin. Kaempferol and quercetin occurred as their 3-O-glucosides and 3-O-galactosides while quercetin also occurred as the 3-O-rhamnoside in *D. perfoliatum*. The diglycoside fraction consisted of the 3-O-rutinosides, 3-O-rhamnosylgalactosides and 3-O-diglucosides of both kaempferol and quercetin. These diglycosides were observed in all specimens taken from the field; they were apparently absent from greenhouse material.

A set of O-methylated compounds based upon quercetin and quercetagetin was also found. 3-O-Methylquercetagetin 7-O-glucoside and 3,3'-di-Omethylquercetin 7-O-glucoside were found in all species, although only a trace of the monomethyl ether was seen in the greenhouse-grown sample of D. perfoliatum. Quercetagetin 7-O-glucoside was observed as a major component of greenhouse D. perfoliatum and as a trace constituent of D. hondurense. 3'-O-Methylquercetagetin 7-O-glucoside was observed only in the cultivated D. perfoliatum. 6-O-Methylquercetagetin (patuletin) 7-Oglucoside was observed in all samples although there was some doubt about its presence in D. perfoliatum collected from nature. 3,6,3'-Tri-O-methylquercetagetin 7-Oglucoside was seen in all specimens except cultivated D. perfoliatum.

In the case of *D. perfoliatum* two samples of plant material were examined, one collected in the field and the other grown from seed obtained from the same collection. These differed in their flavonoid patterns as described above. Four collections of *D. hondurense* and six collections of *D. fruticosum* showed essentially identical 2-dimensional thin layer chromatograms, at least with regard to their flavonoid patterns.

## DISCUSSION

Although only three of the ten species of Desmanthodium have been examined for flavonoids, these taxa are diverse morphologically and represent a reasonable sampling of the genus. Six and four populations of D. fruticosum and D. hondurense, but only one of D. perfoliatum, have been analyzed, respectively.

The results, summarized in Table 1, indicate that the occurrence of flavonoids within each species of Desmanthodium is uniform. This contrasts with the situation in the related genus, Clibadium, in which much infraspecific chemical variation is known [2, 3]. The only variation of this type in Desmanthodium is between fieldand greenhouse-grown material of D. perfoliatum. Seeds from the field population were collected and one plant survived in the greenhouse (not yet having flowered), from which leaves were taken for flavonoid analysis. It is difficult to interpret presence or absence of compounds in the greenhouse material because the genetic composition of this individual is presumably different from that of plants collected in the field (the breeding system is not known). The nature of the growth conditions in the greenhouse, including light quality and intensity, could influence the production of flavonoids [4]. As a result the concentrations of several flavonoids and their glycosides may have dropped below detectable levels, whereas those of quercetagetin 7-O-glucoside and quercetagetin-3'methyl ether 7-O-glucoside have increased.

The distribution of flavonoids among the three species of *Desmanthodium* is essentially the same (Table 1). A trace of quercetagetin 7-O-glucoside occurs in *D. hondurense* and not the others, but this is also found in the greenhouse

Table 1. The major flavonoids of three species of Desmanthodium

Flavonol	Presence/absence in*			
	D. fruticosum	D. hondurense	D. Field	perfoliatum Greenhouse
Kaempferol glycosides				
3-O-Galactoside	+	+	+	n.d.
3-O-Glucoside	+-	+	+	+
3-O-Gal-rhamnoside	+	+	+	n.d.
3-O-Rutinoside	+	+	+	*****
3-O-Diglucoside	+	+	+	****
Quercetin glycosides				
3-O-Galactoside	+	+	+	n.d.
3-O-Glucoside	+	+	+	+
3-O-Rhamnoside	_		+	
3-O-Rutinoside	+	+	+	tr
3-O-Diglucoside	+	+	+	tr
3-O-Rhamnoside	+	+	+	n.d.
Methylated derivatives				
Patuletin-3-O-glycoside	+	+	?	+
3-Methylquercetagetin 7-O-glucoside	+	+	+	tr
3,3'-Dimethylquercetin 7-O-glucoside	+	+	+	+
3,3'6-Trimethylquercetagetin 7-O-glucoside	+	+	+	
Quercetagetin 7-O-glycoside	ween	tr	-	+
Quercetagetin 3'-methyl ether 7-O-glycoside			_	+

<sup>\*</sup>n.d. = not determined, tr = trace. Numbers of populations analysed were 6, 4, 1 and 1, respectively.

material of *D. perfoliatum* (not from the field-collected material, however). Also, *D. perfoliatum* (field) has traces of patuletin 3-O-glucoside in contrast to the other two species, but greenhouse material of the same taxon showed detectable quantities.

The close morphological relationship Desmanthodium to Clibadium [1] is paralleled by the similarity of flavonoids. In both genera, kaempferol and quercetin occur with several methylated derivatives. Both contain quercetagetin and accumulate derivatives methylated at the 3'- and 6-positions. Desmanthodium is different in accumulating mono-methyl derivatives at position 3, quercetin dimethyl ethers and trimethyl compounds. By contrast, Clibadium accumulates directly derivatives of quercetagetin and Desmanthodium lows not. However, these compounds would be expected to occur in the latter genus, probably at low (and perhaps undetectable) concentrations because they are probably biosynthetic intermediates.

Morphologically, Clibadium is closest to Ichthyothere, a genus of about 15 species primarily from northeastern South America, but for which no flavonoid data are presently available. (The two genera are quite similar, however, in their polyacetylenes [2,5].) Desmanthodium relates to Clibadium somewhat more distantly than the latter does to Ichthyothere. The meager chromosomal data support this interpretation. Clibadium is n = 16 and 24 (probably x = 8) [6-12], and Ichthyothere is n = ca 33 (probably ca 32, also x = 8) [13], but Desmanthodium is different with n = 18 [14,15], with probably x = 9. Recently Robinson described two new subtribes of the Heliantheae [16]: Clibadiinae and Desmanthodiinae. Although details of these new subtribes have not yet been

provided, such as the genera included other than the obvious types, it seems unlikely that *Clibadium* and *Desmanthodium* are so distantly related that they deserve to be separated at the subtribal level. Certainly the flavonoid data provide no support for such a view.

### **EXPERIMENTAL**

Source of plants (All Stuessy and Gardner are at OS; asterisks indicate collections from which the initial analysis for flavonoids was done.) Desmanthodium fruticosum Greenm.: Mexico: Guerrero, 4206, 4215; Jalisco, 4135, 4147; Michoacán, 4173; Nayarit, 4115.\* D. hondurense Molina: Honduras: Comayagua, 4408,\* 4414; Ocotopeque, 4390, 4401. D. perfoliatum Benth.: Mexico: Chiapas, 4306\*.

Extraction and isolation procedures. Plant material was extracted repeatedly with boiling 80% MeOH. Combined extracts were evapd to dryness in vacuo, the residue re-extracted with hot water and filtered using a diatomaceous filter aid. Extraction of the aq. fraction with water-saturated n-BuOH afforded the polyphenolic fraction. After removal of n-BuOH under red. pres., the residue was taken up in MeOH for storage. Subsequent fractionations were accomplished using column chromatography on Sephadex LH-20 followed by partition separations, all as described by Wilkins and Bohm [17]. Final purifications were accomplished by prep. TLC [17].

Identification of compounds. The mono- and diglycosides of kaempferol and quercetin were identified by routine methods including UV analysis [18], partial and total hydrolyses and comparison with known compounds. In the diglycosides, the comparison standards were the corresponding compounds obtained from Clibadium [2]. Quercetagetin 7-O-glucoside, 6-O-methylquercetagetin (patuletin) 7-O-glucoside, and 3'-O-

methylquercetagetin 7-O-glucoside were identified by comparison with compounds isolated from Clibadium species [2].

The structure of 3-O-methylquercetagetin 7-O-glucoside was arrived at in the following way. UV characteristics suggested it to be a flavonol. Acid hydrolysis was very slow under conditions which cleave 3-O-glycosides readily; the sugar released was glucose. That it was a flavonol 7-O-glucoside was confirmed by the absence of an acetate-induced shift of band II. Acetate did cause a significant bathochromic shift of band I and addition of borate indicated the presence of 3',4'-dihydroxylation. This agreed with the yellow-orange colour reaction observed with diphenylboric acid ethanolamine complex which is characteristic of the quercetin-type B-ring. Addition of AlCl<sub>3</sub> yielded a band I shift of 87 nm; addition of HCl resulted in a band I bathechromic shift of 19 nm relative to the MeOH spectrum. This behaviour indicates the presence of 6-oxygenation plus a substituted 3hydroxyl function (either sugar or methyl) [18]. Both the glycoside and the aglycone appeared dark under UV light which requires that the 3-position be substituted. Since there is a sugar at the 7-position, the methyl group must be at position 3. TLC against a series of methylated standards suggested that the unknown aglycone bears a single O-methyl function. Thus, the compound must be 3-O-methylquercetagetin 7-O-glucoside.

The structural determination of 3,3'-di-O-methylquercetin 7-O-glucoside proceeded as follows. The compound exhibited UV characteristics of a flavonol and appeared as a dark absorbing spot on chromatograms. Acid hydrolysis, which yielded glucose, was very slow indicating glycosylation at the 7-position. The aglycone also appeared dark, suggesting that the 3-position was still substituted. UV spectral properties suggested the presence of 3',4'-oxygenation but the absence of both a borate shift and the typical quercetin yellow-orange colour with the diphenylborate spray argued for substitution at one of these positions. Acetate did not alter the band II position of the glycoside but did produce a band I shift suggesting an unsubstituted 4'-hydroxyl. AlCl<sub>3</sub> gave a 56 nm bathochromic shift of band I which was stable with acid; therefore, 6-oxygenation could be eliminated. TLC against methylated standards suggested the presence of two O-methyl functions. The compound is thus 3,3'-di-O-methylquercetin 7-Oglucoside.

Finally, the trimethylated flavonoid was determined by the following means. This compound also exhibited the UV and hydrolytic characteristics of a flavonol 7-O-glucoside having 3',4'-oxygenation and oxygenation at position 6. The compound was dark under UV before and after hydrolysis indicating the 3-

O-methyl group. The <sup>1</sup>H NMR spectrum of the glucoside showed the quercetagetin pattern (H-8, H-2', H-5' and H-6') plus three O-methyl groups. UV spectral data agreed with those published [18] for jacein, 3,6,3'-tri-O-methylquercetagetin 7-O-glucoside.

Acknowledgements—Appreciation goes to Robert Gardner for help in the field; the National Science Foundation (grant DEB-7520819) for financial support to T.F.S.; and the NRCC (now NSERC) for operating and equipment grants to B.A.B. Dr. Elijah Tannen made his usual contribution, for which we express our thanks.

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